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1 1st Year Chemistry Mcqs Solved For Fsc, MCAT, ECAT Entry Tests Practice

The number of moles of CO_2 which contain 8.0 g of oxygen: (a) 0.25 (b) 0.50 (c) 1.0 (d) 1.50

Empirical formula of Benzene is: (a) CH_2O (b) CHO (c) C_6H_6 (d) C_6H_2

The comparative rates at which the solutes move in paper chromatography depends on: (a) The size of paper (b) R_f values of solution (c) Temperature of the experiment (d) size of the Chromatographic tank/leaf

In 1873, Plasma was identified by the scientist: (a) John Dalton (b) Chadwick (c) William Crookes (d) Soddy

Ice occupies more space than liquid water: (a) 5% (b) 10% (c) 15% (d) 12%

Which of the given is a Pseudo Solid: (a) CaF_2 (b) Glass (c) NaCl (d) All these

Quantum number value for 2p subshell are: (a) $n=2, l=1$ (b) $n=1, l=2$ (c) $n=1, l=0$ (d) $n=2, l=0$

Mass of Electron is: (a) $9.1095 \times 10^{-31} \text{ Kg}$ (b) $9.1095 \times 10^{-31} \text{ Kg}$ (c) $9.1095 \times 10^{-27} \text{ Kg}$ (d) $9.1095 \times 10^{-31} \text{ g}$

The number of bonds in Nitrogen Molecule is:

(a) One sigma and one pi bond (b) One sigma and two pi (c) Three sigma only (d) Two sigma and one pi

Noble gases are highly stable and least reactive because:

(a) They are gases (b) their valence shell are complete (c) They are present in zero group (d) They are very safe

Burning of coal is the example of:

(a) Non spontaneous Reaction (b) Reversible Reaction (c) Spontaneous Reaction (d) Irreversible Reaction

For which system does the equilibrium constant, K_c has units of (concentration) $^{-1}$:

(a) $N_2 + 3H_2 \rightleftharpoons 2NH_3$ (b) $H_2 + I_2 \rightleftharpoons 2HI$ (c) $2NO_2 \rightleftharpoons N_2O_4$ (d) $2HF \rightleftharpoons H_2 + F_2$

The relationship between K_p and K_c is given by:

(a) $K_p = K_c (P)^{-\Delta n}$ (b) $K_p = K_c (P)^{\Delta n}$ (c) $K_p = K_c (RT)^{\Delta n}$ (d) $K_p = K_c (RT)^{-\Delta n}$

The molal boiling point constant is the ratio of the elevation in boiling point to:

(a) Molality (b) Molarity (c) Mole fraction of solvent (d) Mole Fraction of Solute

If the Salt Bridge is not used between two half cells, then the voltage:

(a) Decrease Rapidly (b) Decrease Slowly (c) Does not change (d) Drops to zero

The electrolyte used in fuel cell is:

(a) Aqueous NaCl (b) Molten NaCl (c) KOH (d) $NaNO_3$

A substance which promotes the activity of a catalyst is called:

(a) Activator (b) Inhibitor (c) Retarder (d) Auto catalyst

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