

# Download File PDF Oxidation Number Practice Worksheet Answers

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## Quiz & Worksheet - How to Assign Oxidation Numbers to Elements in a Chemical Formula

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1. What are the oxidation numbers of the atoms in the compound  $\text{CF}_4$ ?

- a. +4, 4
- b. +4, -4
- c. -4, 4
- d. -4, -4

2. What is the oxidation number of each atom in  $\text{KH}$ ?

- a. +1, 1
- b. -1, 1
- c. -1, -1
- d. +1, -1

3. What is an oxidation number?

- a. A number assigned to atoms in molecules to show the exact distribution of the electrons.
- b. A number assigned to atoms in molecules to show the general distribution of the electrons.
- c. A number that tells you exactly which atom has how many electrons at any given time.
- d. A number assigned to atoms in molecules to show the exact distribution of the protons.
- e. A number assigned to atoms in molecules to show the general distribution of the protons.



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